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1 **Interspecies comparison of plasma metabolism and sample stabilization for quantitative**  
2 **bioanalyses: application to (*R*)-CE3F4 in preclinical development, including metabolite**  
3 **identification by high-resolution mass spectrometry**

4

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1 **Abstract**

2 The CE3F4 is an inhibitor of the type 1 exchange protein directly activated by cAMP  
3 (EPAC1), which is involved in numerous signaling pathways. The inhibition of EPAC1  
4 shows promising results *in vitro* and *in vivo* in different cardiac pathological situations like  
5 hypertrophic signaling, contributing to heart failure, or arrhythmia. An HPLC-UV method  
6 with a simple and fast sample treatment allowed the quantification of (*R*)-CE3F4. Sample  
7 treatment consisted of simple protein precipitation with 50  $\mu$ L of ethanol and 150  $\mu$ L of  
8 acetonitrile for a 50  $\mu$ L biological sample. Two wavelengths were used according to the  
9 origin of plasma (220 or 250 nm for human samples and 250 nm for murine samples).  
10 Accuracy profile was evaluated for both wavelengths, and the method was in agreement with  
11 the criteria given by the EMA in the guideline for bioanalytical method validation for human  
12 and mouse plasma samples. The run time was 12 min allowing the detection of the (*R*)-CE3F4  
13 and a metabolite. This study further permitted understanding the behavior of CE3F4 in plasma  
14 by highlighting an important difference between humans and rodents on plasma metabolism  
15 and may impact future *in vivo* studies related to this molecule and translation of results  
16 between animal models and humans. Using paraoxon as a metabolism inhibitor was crucial  
17 for the stabilization of (*R*)-CE3F4 in murine samples. HPLC-UV and HPLC-MS/MS studies  
18 were conducted to confirm metabolite structure and consequently, the main metabolic  
19 pathway in murine plasma.

20

21 **Keywords:** EPAC1, High-performance liquid chromatography, Species metabolism,  
22 Validation method, Plasma stability study

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## 1 **1 Introduction**

2 Heart disease is one of the major issues of the last century which burden remains significant  
3 with high morbidity and mortality that has not been improved by any treatment modality [1].  
4 Many efforts are performed to find new efficient drugs in the cardiology field. Very recently,  
5 one *in vivo* study described the potential of a new drug, CE3F4, on C57BL/6 mouse models  
6 for atrial/ventricular arrhythmias [2]. CE3F4 is a tetrahydroquinoline analog identified as a  
7 specific and uncompetitive inhibitor of EPAC1 (exchange protein directly activated by cAMP  
8 1) [3,4]. The downregulation of EPAC1 activity can restore or protect physiological  
9 functions, which could be exploited for new treatments of heart failure [5] and arrhythmia  
10 [2,6]. CE3F4 has demonstrated *in vitro* efficient inhibition of cellular mechanisms involving  
11 EPAC1 in pathological cardiac conditions [7–9]. Moreover, other diseases such as breast,  
12 gastric, and pancreatic cancers [10–12], Alzheimer’s disease [13], and depression [14] may  
13 benefit from CE3F4. Altogether, these results were promising for further pharmaceutical  
14 development of this drug.

15 Extensive and comprehensive preclinical studies are required to yield toxicological,  
16 pharmacokinetic, and pharmacological information before clinical trials of a new drug [15].  
17 Typically, pharmacokinetic/pharmacodynamic (PK/PD) study involves *in vitro* analysis from  
18 animals (e.g. rodents) and human plasma [16]. Quantitative analysis is of utmost importance  
19 for this purpose [16,17]. However, particular care should be taken in analytical strategies due  
20 to metabolism differences between species. To facilitate the preclinical results to clinical  
21 study transposition, drug metabolism should be performed *in vitro* with murine and human  
22 plasma at an early stage of development. This raises the challenge of the analysis in a  
23 complex medium such as plasma, containing various degrading enzymes that differ from  
24 species. Indeed, it is noteworthy that differences in the type of enzyme existing between  
25 mammalian species may impact drug analysis and subsequently lead to considerable  
26 challenges in interpreting the data. Interspecies differences in drug metabolism were reported  
27 between rodents and humans [18–20]. For example, carboxylesterases, which are known to  
28 degrade drugs containing amides, carbamate, esters, or thioesters bonds [21,22], are present in  
29 rodent plasma but not in human plasma [23]. Therefore, plasma metabolism can lead to drug  
30 degradation during sample collection, preparation, or storage [24], inducing a false estimate  
31 of the drug level in plasma which is detrimental for monitoring purposes [25]. To overcome  
32 these issues, several procedures are currently used for *ex vivo* drug stabilization before

1 analysis. To achieve drug stabilization in plasma samples, specific enzyme inhibitors,  
2 involved in the degradation, have been described such as irreversible non-competitive  
3 inhibitors like pesticides inhibiting esterases to stabilize diformyldapsone [19], oseltamivir  
4 [25], and zeylenone [26]. Other inhibitors like tetrahydrouridine demonstrated an interest in  
5 gemcitabine stabilization (competitive inhibitor) before quantification [27], or EDTA  
6 (ethylenediaminetetraacetic acid) which can be used to inhibit metal-dependent enzymes  
7 [28,29]. Less-specific methods can be used such as cold temperature to slow-down enzymatic  
8 reaction kinetic, pH setting outside optimal range for enzyme activity [30], or protein  
9 precipitation by organic solvent [31,32], this latest method presenting the advantage to  
10 deproteinize the plasma sample making it suitable for analysis. Very recently, different  
11 methods were combined to stabilize molecules of interest in plasma before analysis, as EDTA  
12 with sample acidification and refrigerated conditions for osimertinib stabilization [33], or  
13 sodium fluoride (esterase inhibitor) associated to sample acidification, EDTA, derivatization,  
14 and refrigerated conditions for the simultaneous quantification of vicagrel and its two  
15 metabolites [34].

16 Nevertheless, when it comes to a new drug, the enzymes likely to affect the clearance of the  
17 drug are not always identified. Di *et al.* [35] emphasized the importance of plasma stability  
18 experiments on hydrolyzable drugs during drug discovery, as this type of study is very  
19 informative about compound screening, drug liability, and the potential need to use hydrolase  
20 inhibitors during PK studies [35]. In the case of (*R*)-CE3F4, to the best of our knowledge, no  
21 degrading enzyme in plasma has yet been reported. To conduct PK/PD studies, the behavior  
22 of (*R*)-CE3F4 in blood or plasma medium (e.g. differences in pharmacokinetics, metabolism)  
23 should be conducted [36,37]. CE3F4 shows a N-formamide function (**Fig. 1**), and thus, has  
24 the potential to be inactivated by hydrolases, including esterases or N-formamidase [18–20].  
25 Due to the known differences between humans and rodents in the expression of the wide class  
26 of hydrolases [23,38,39], a specific approach must be considered.

27 The present paper describes the first method for (*R*)-CE3F4 quantification in both human and  
28 mouse plasma using LC-UV system. The method included enzyme inhibition, extraction, and  
29 assay steps. The method was validated through the European Medicine Agency guideline for  
30 bioanalytical analysis, on two wavelengths (220 and 250 nm) to overcome a lack of  
31 specificity depending on the origin of the plasma matrices analyzed (i.e. human or mouse). It

1 was then applied to plasma samples, and an essential difference in plasma metabolism has  
2 been highlighted *ex vivo* between murine and human plasma.

## 3 **2 Materials and methods**

### 4 *2.1 Chemicals*

5 A racemic mixture of CE3F4 and the deformed-CE3F4 were synthesized using published  
6 procedures [40]. The active enantiomer (*R*)-CE3F4 was isolated with a purity >99% and  
7 enantiomeric excess (ee) >98% by applying the racemic mixture to a 250 × 30 mm  
8 Chiralpak® IB column (Daicel Corporation, Japan) equilibrated with n-hexane/propan-2-ol  
9 (95/5) at 42 mL/min. Paraoxon-ethyl (paraoxon) with the standard analytical grade,  
10 ethylenediaminetetraacetic acid disodium salt dihydrate (EDTA) with purity >99%, benzil  
11 with purity >98%, and bis(4-nitrophenyl) phosphate (BNPP) with purity >99%, polyethylene  
12 glycol 200 Da (PEG<sub>200</sub>), Solutol® HS 15, phosphate-buffered saline (PBS) at pH 7.4 (sterile  
13 and suitable for cell culture) were acquired from Sigma-Aldrich (Saint-Louis, MO, USA). L-  
14 α-phosphatidylcholine from egg (EggPC), 1,2-distearoyl-sn-glycero-3-phosphoethanolamine-  
15 N-[methoxy(polyethyleneglycol)-2000] (DSPE-PEG<sub>2000</sub>) were obtained from Avanti.  
16 Miglyol® 812 was purchased from Sasol (Johannesburg, South Africa). Zoletil® 50  
17 (tiletamine, zolazepam), Rompun® 2% (Xylazine), Iso-Vet (isoflurane) were respectively  
18 purchased from Virbac (Carros, France), Bayer (Leverkusen, NRW, Germany), and Piramal  
19 Critical Care (Voorschoten, Netherlands). Ethanol (EtOH), as well as dimethyl sulfoxide  
20 (DMSO), were bought from VWR (Radnor, PA, USA), acetonitrile (ACN) with HPLC grade  
21 was purchased from Carlo Erba Reagents (Val-de-Reuil, France). Deionized water (H<sub>2</sub>O) was  
22 produced by a Milli-Q water purifying system (Millipore, Milford, MA, USA). For the LC-  
23 MS experiment, water, acetonitrile, and formic acid were acquired from Merck KGaA  
24 (Darmstadt, Germany).

### 25 *2.2 Biological samples*

26 Human blank plasmas were obtained from blood donations from healthy volunteers who  
27 signed written informed consent (French Blood Establishment, La Plaine Saint-Denis,  
28 France). A sampling of murine plasmas was carried out according to the ethical committee  
29 (agreement number APAFIS#5584-20 16050312271349 v4). Blood samples were collected in  
30 heparinized tubes. After centrifugation at 500 g for 10 minutes in polypropylene tubes,  
31 plasmas were separated from the pellets and stored at -20°C.

### 32 *2.3 Preparation of (R)-CE3F4 stock solutions, calibration standards, and quality controls*

1 All liquids were handled with MICROMAN E positive displacement pipettes (Middleton, WI,  
2 USA) due to the viscosity of plasma samples and the use of organic solvents. Two stock  
3 solutions of (*R*)-CE3F4 were prepared in ethanol at a concentration of 1000 µg/mL, for  
4 calibration, and quality controls (QC). They were stored at +5°C. For the preparation of  
5 calibration standards, serial dilution (dilution factor = 2.43) in ethanol was performed from  
6 the stock solution to obtain an eight levels calibration curve with concentrations ranging from  
7 0.40 to 200 µg/mL (**Table 1**). For the preparation of QC, appropriate volumes of stock  
8 solutions were evaporated to dryness at 40°C under a stream of nitrogen in 10 mL calibrated  
9 flasks for human QC or in 5 mL calibrated flasks for murine QC. For human QC, the dry  
10 residue was taken up with 10 mL of blank human plasma and mixed under magnetic stirring  
11 for two hours to obtain appropriate solubilization of the residue in the plasma at the six  
12 concentrations levels from 0.40 to 150 µg/mL (**Table 1**). For murine QC, 5 mL of mouse  
13 plasma loaded with paraoxon (final concentration of 10<sup>-4</sup>M) were used to obtain three  
14 concentration levels of (*R*)-C3F4 (0.40, 8.00, and 150 µg/mL) (**Table 1**). These QC were  
15 defined as working solutions and were stored at -20°C.

#### 16 *2.4 Sample treatment*

17 Using 2 mL polypropylene microcentrifuge tubes, 50 µL of calibration standards were added  
18 to 50 µL of (*R*)-CE3F4-free human plasma for the calibration samples and followed by a  
19 vortex-mixing step (10 seconds). For plasma samples spiked with (*R*)-CE3F4 (human plasma  
20 as QC for validation, human and murine plasmas for metabolism experiments), 50 µL of  
21 plasma was supplied with 50 µL of ethanol and followed by a vortex-mixing step (10  
22 seconds). The proportions for each component (plasma, ethanol) were identical between  
23 calibration samples and QC (**Table 2**). Protein precipitation was performed by the addition of  
24 150 µL of acetonitrile followed by a vortex-mixing for 10 seconds. After precipitation, tubes  
25 were centrifuged at 13,000 g for 10 minutes at room temperature, then 100 µL of the  
26 supernatant was collected into autosampler vials. 15 µL of the supernatant was injected into  
27 the chromatographic system.

#### 28 *2.5 Stock solutions of enzyme inhibitors*

29 Several stock solutions of paraoxon, BNPP, and benzil were separately prepared at the  
30 concentration of 10<sup>-2</sup> M, 10<sup>-3</sup> M, and 10<sup>-4</sup> M in DMSO. EDTA was solubilized in water to  
31 obtain a concentration of 200 mM.

#### 32 *2.6 Instruments*

### 1 2.6.1 *UV/Visible spectrophotometer*

2 The UV absorbance spectra of (*R*)-CE3F4 were acquired using a Lambda 25 UV/VIS  
3 spectrometer from PerkinElmer (Waltham, MA, USA). The acquisition was performed with  
4 the software (UV WinLab from PerkinElmer) in scan mode from 195 to 400 nm with an  
5 acquisition rate of 480 nm/min, a slit width of 1 nm, and a data interval of 1 nm. Samples  
6 were analysed in a quartz cuvette (Hellma Analytics, Paris, France), and the optical  
7 pathlength was 10 mm. Appropriate dilutions of (*R*)-CE3F4 in ACN were performed to obtain  
8 4.0, 8.0, and 16.0 µg/mL. After subtraction of the blank (ACN), absorbances at each  
9 wavelength have been divided by the corresponding concentration of the analysed sample to  
10 obtain the molar extinction coefficient.

11 Second derivate UV spectra were calculated using the Savitzky-Golay (SG) algorithm to  
12 identify UV maxima. Such a procedure highlights the potential wavelengths of interest for  
13 setting up chromatographic analysis. The convolution coefficients used were those to perform  
14 the SG second derivative with a quadratic/cubic polynomial fitting and a window size of nine  
15 points.

### 16 2.6.2 *Liquid chromatography with UV detection*

17 LC was performed on a Waters HPLC system (Milford, MA, USA) equipped with a diode  
18 array detector (Waters 2996 PDA), an injector (Waters 717 plus Autosampler), and a  
19 quaternary pump (Waters 600 HPLC pump). An external heater/chiller for the column was  
20 used (GRACE Model 7956R, Columbia, MD, USA). The separation was performed on  
21 XBridge C18 3.5 µm, 4.6 × 100 mm column (Waters). The oven temperature was set at 35°C.  
22 The mobile phase consisted of a mixture of water and acetonitrile (40:60, v/v) under isocratic  
23 mode at a flow rate of 1 ml/min. The acquisition was carried out from 194 to 400 nm with a  
24 resolution of 1.2 nm. Chromatograms were registered at 220 and 250 nm.

### 25 2.6.3 *Liquid chromatography with MS/MS detection*

26 Analysis was performed on RP-HPLC (ACQUITY, Waters, Manchester, UK) hyphenated to  
27 LTQ-orbitrap XL mass spectrometer (Thermo-Fischer Scientific, Les Ulis, France)  
28 specifically for identification purpose. The chromatographic conditions were very similar to  
29 the HPLC-UV method. Differences stood in the addition of formic acid 0.1% (v/v) into the  
30 mobile phase (water and ACN, 40:60, v/v) and in the flow rate which was set at 0.5 mL/min.  
31 The column remaining the same (XBridge C18 3.5 µm, 4.6 × 100 mm) was kept at the same  
32 temperature (35°C). Mass spectrometry analyses were conducted with a heated electrospray



1 ionization (ESI) interface (HESI-II) operating on positive mode with the following  
2 parameters: ESI voltage of +4.0 kV, sheath gas flow rate value of 80, and an auxiliary gas  
3 flow rate value of 12. Source heating temperature was set to 300°C, capillary and tube lens  
4 voltage to a value of 35 V and 90 V respectively. Mass/charge (m/z) range was 50-500. Full  
5 MS scans were recorded in profile mode using the high-resolution FTMS analyzer (R =  
6 60,000). MS/MS fragmentation was performed by collision-induced dissociation (CID) using  
7 a normalized CID energy of 35. Data acquisition was controlled by Xcalibur software  
8 (Thermo-Fischer Scientific, Les Ulis, France).

### 9 *2.7 Analytical method validation for (R)-CE3F4 plasma stability studies*

10 The validation procedure was adapted from the European Medicine Agency (EMA) guideline  
11 for bioanalytical method validation which defines the acceptance criteria [41]. This procedure  
12 was carried out with plasma samples on three separate days, following the methodology from  
13 the *Société Française des Sciences et Techniques Pharmaceutiques* (SFSTP) previously  
14 reported [42–44]. LC-UV system has been used, and the signal was the chromatographic peak  
15 area at 220 nm and 250 nm.

#### 16 *2.7.1 Specificity*

17 The specificity of this method was demonstrated by injecting the extracts blank plasma from 6  
18 human or murine sources, the extracts of plasma spiked with other compounds like enzyme  
19 inhibitors (i.e. paraoxon, BNPP, benzil), or molecules which can be used for *in vivo*  
20 experiment as isoflurane, or the association Zoletil® (tiletamine, zolazepam) plus Rompun®  
21 (xylazine) at lethal dose. The specificity of the method was also explored regarding excipients  
22 which can be used for the formulation of the (R)-CE3F4 (PEG<sub>200</sub>, Solutol HS 15, EggPC,  
23 DSPE-PEG<sub>2000</sub>, Miglyol 812). No peak with a surface area equal to at least 20 % of the LLOQ  
24 should appear on the chromatogram under or near the (R)-CE3F4 (resolution  $R_s < 1.5$ ).

#### 25 *2.7.2 Quantitative validation*

26 In our experiments, we assumed that a calibration curve and its corresponding QCs were  
27 accepted if 6 out of 8 CS concentration levels and, at least 4 out of 5 QC samples for each QC  
28 level (= 80% of the QCs samples, corresponding to the  $\beta$ -expectation tolerance interval  
29 described by the SFSTP [42–44]) satisfied the validation criteria. This approach is a  
30 derivation of EMA guidelines, which recommend that 67% of QC samples for 50% QC level  
31 must fulfill the bias acceptance criteria.

1 For the validation of the quantitative method, the calibration curve was fitted by least square  
2 regression using 8 CS concentration levels, ranging from 0.40 µg/mL to 200 µg/mL, and  
3 different weighting factors were evaluated (1, 1/concentration, 1/concentration<sup>2</sup>). Linearity  
4 was assessed from the slope of the linear regression and the y-intercept and the coefficient of  
5 determination (R<sup>2</sup>). For the validation of the method, accuracy, precision, and recovery were  
6 tested with the same QC sample. Accuracy, repeatability, and intermediate precision were  
7 assessed using five determinations with independent extractions of the same batch, on three  
8 separate days (n = 15) for each of the six QC levels using human plasma, and each of the  
9 three QC levels using murine plasma loaded with paraoxon. Accuracy was calculated by the  
10 mean bias between the observed back-calculated concentration (C<sub>measured</sub>) and the theoretical  
11 concentration (C<sub>theoretical</sub>), i.e.  $\text{Bias}(\%) = 100 \times (C_{\text{measured}} - C_{\text{theoretical}}) / C_{\text{theoretical}}$ . Within-assay  
12 accuracy was calculated for each day using the 5 available values, and within-assay precision  
13 was determined for each QC level from its mean back-calculated concentration and mean  
14 variance (i.e.  $\text{coefficient of variation}(\%) = 100 \times \text{standard deviation} / \text{mean value}$ ). Cochran test  
15 ( $\alpha = 0.05$ ) was used to check homogeneity of variances over the 3 days before using the  
16 mean variance. Between-assay precision and accuracy were calculated for the 6 QC levels on  
17 3 values (i.e. one value per day). For each day, the selected value was the one presenting the  
18 most important bias. For each QC level, between-days precision and accuracy were  
19 respectively the coefficient of variation and the mean bias calculated on these three selected  
20 samples respectively. Both precision and accuracy were combined as the tolerance interval to  
21 build the accuracy profile. The lower limit of quantification (LLOQ) and the upper limit of  
22 quantification (ULOQ) of the method were both set to the lowest and the highest QC value  
23 (0.40 µg/mL and 150 µg/mL), which is another adaptation of EMA guidelines. Mean relative  
24 errors (%) from the expected concentrations were calculated and the accepted ranges were  $\pm$   
25 20 % and  $\pm$  15 % for the lowest QC concentration (i.e. 0.40 µg/mL), and the other QC,  
26 respectively. Acceptance criteria limits for the  $\beta$ -expectation tolerance interval were  $100 \pm$   
27 20% for the first QC level (LLOQ), and  $100 \pm 15\%$  for other QC levels. The carryover was  
28 also investigated with the injection of the highest calibration standard (200 µg/mL), followed  
29 by five injections of blank plasma extract. Carryover was considered negligible if the signal  
30 obtained with the 1<sup>st</sup> blank following the highest standard was less than 20% of the LLOQ.

### 31 2.7.3 *Extraction recovery and stability*

1 Two calibration curves were carried out, the difference was in the moment of the (*R*)-CE3F4  
2 addition. The first calibration curve corresponded to addition into plasma before the  
3 extraction process (calibration “A”), and the second calibration curve consisted in the addition  
4 of the (*R*)-CE3F4 after the extraction process on blank plasma with ACN, and it corresponds  
5 to a theoretical recovery of 100 % (calibration “B”) [45]. Samples were injected into the  
6 chromatographic system. The recovery was calculated each day of the validation process with  
7 the ratio between the slopes of these 2 calibration curves (A/B).

8 The stability of the extracts in the autosampler (set at 20° C) was investigated over a 15 h  
9 period by comparing the concentration obtained with extracts divided into two batches. One  
10 was injected immediately after extraction then reinjected after 15 h in the autosampler, and  
11 the other batch was just injected once after 15 h in the autosampler. The stability over one  
12 year was investigated for stock solutions for their usual storage conditions (5° C), one month  
13 for human QCs or one week for mouse QCs stored at -20° C, after 3 cycles of freeze-thaw (-  
14 20° C to ambient temperature), and short-term stability for two hours at room temperature. If  
15 a decrease of 15 % or more was observed for QC or 5 % for stock solutions, the analyte was  
16 considered unstable for the corresponding tested condition.

## 17 2.8 Evaluation of (*R*)-CE3F4 ex vivo stability in human or murine plasma

### 18 2.8.1 (*R*)-CE3F4 behaviour in plasma at 37°C

19 The stability of (*R*)-CE3F4 in human and murine plasmas was investigated ex vivo for 24  
20 hours at 37°C. Due to the poor solubility in water media, (*R*)-CE3F4 was first solubilized into  
21 DMSO to obtain a stock solution at 10 mg/mL. 3.6 µL of the (*R*)-CE3F4 solution was mixed  
22 to 600 µL plasma in microcentrifuge tubes to obtain a concentration near 60 µg/mL.  
23 Microcentrifuge tubes were put in a rotary agitator at 37°C. Sampling was achieved at the  
24 following times ( $t = 5', 15', 30', 45', 60', 90', 120', 180', 20 \text{ h and } 24 \text{ h}$ ). To follow the stability  
25 of the (*R*)-CE3F4 over time, a normalization of the data was made by the ratio between the  
26 concentration measured at each sampling time and the concentration measured at 5 minutes  
27 ( $[\text{CE3F4}]_t / [\text{CE3F4}]_{5\text{min}} = \text{relative concentration (\%)} \text{ at time } t$ ). Experiments were repeated  
28 four times with human plasma, and six times with murine plasma.

### 29 2.8.2 Stabilization studies of (*R*)-CE3F4 in murine plasma at 37°C with EDTA, paraoxon, 30 BNPP, or benzil

31 Paraoxon, BNPP, benzil, and EDTA were separately tested as metabolism inhibitors in  
32 murine plasma to stop the degradation of (*R*)-CE3F4 that may occur during the processing of

1 samples *in vivo* studies. EDTA was tested at the concentration of 10 mM in murine plasma  
2 sample (i.e. 15  $\mu\text{L}$  of the EDTA stock solution were added to 284  $\mu\text{L}$  of plasma) spiked with  
3 (*R*)-CE3F4 (i.e. 1.8  $\mu\text{L}$  of (*R*)-CE3F4 at 10 mg/mL for 300  $\mu\text{L}$  of plasma) and the sample was  
4 put in a rotary agitator at 37°C after one-minute vortex step. Paraoxon, BNPP, and benzil  
5 were studied separately at  $10^{-4}$  M,  $10^{-5}$  M,  $10^{-6}$  M in plasma. Briefly, for these three inhibitors,  
6 3.0  $\mu\text{L}$  of each stock solution (at  $10^{-2}$  M,  $10^{-3}$  M,  $10^{-4}$  M) were added in 296  $\mu\text{L}$  of murine  
7 plasma, followed by a one-minute vortex step and five minutes in a rotary agitator at 37°C.  
8 Then, (*R*)-CE3F4 was added in plasma already loaded with inhibitor (paraoxon or BNPP or  
9 benzil) in the same proportions described in the paragraph above (i.e. 1.8  $\mu\text{L}$  of (*R*)-CE3F4 at  
10 10 mg/mL for 300  $\mu\text{L}$  of plasma). For the four compounds, five sampling times were analysed  
11 for up to 1 hour to estimate the stabilization of the (*R*)-CE3F4 in samples. Similarly, in a  
12 second experiment (*R*)-CE3F4 and paraoxon (stock solution at  $10^{-2}$  M) were simultaneously  
13 added in murine plasma at the same proportion and conditions of the previous experiment to  
14 evaluate whether (*R*)-CE3F4 degradation occurred during homogenization of the sample.

### 15 2.8.3 *Storage conditions suitable for (R)-CE3F4 in murine plasma*

16 The stability of loaded murine plasma samples with paraoxon ( $10^{-4}$  M in plasma sample) and  
17 (*R*)-CE3F4 (60  $\mu\text{g/mL}$ ) was investigated at 5° C for 3 days, after 3 days and one week at -20°  
18 C, and after 3 cycles of freeze-thaw. If a decrease of 15% or more was observed, samples  
19 were considered unstable.

### 20 2.8.4 *HPLC-MS/MS for metabolite identification*

21 The identification of (*R*)-CE3F4 metabolite was performed by HPLC-MS/MS analysis. To  
22 proceed with the identification, a control sample of (*R*)-CE3F4 (5  $\mu\text{g/mL}$ ), a control sample of  
23 deformedylated-CE3F4 (5  $\mu\text{g/mL}$ ), and plasma extract (as described in 2.8.1) were analyzed.

### 24 2.9 *Evaluation of (R)-CE3F4 protection against enzymatic degradation by various* 25 *formulations*

26 Three different formulations were used first to solubilize the (*R*)-CE3F4 to obtain a final  
27 concentration of 625  $\mu\text{g/mL}$ . In the case of DSPE-PEG<sub>2000</sub> micelles, a thin-film hydration  
28 method was used [46]. Briefly, (*R*)-CE3F4 was incorporated simultaneously with DSPE-  
29 PEG<sub>2000</sub> in the chloroform solution, then the mixture was dried at ambient temperature for 60  
30 min using a rotary evaporator R-215 (Büchi, Switzerland) and the obtained film was hydrated  
31 with PBS at room temperature ( $[\text{DSPE-PEG}_{2000}] = 25$  mM,  $[(\text{R})\text{-CE3F4}]_{\text{post process}} = 900$   
32  $\mu\text{g/mL}$ ). In the cas of Solutol/PEG, a mixture of Solutol HS 15, PEG<sub>200</sub>, and PBS (4.5:1:15,

1 mPEG200/mSolutol HS 15/vPBS) was used. For 1.0 mL of preparation, 0.625 mg of (*R*)-CE3F4 was  
2 dissolved in 225 mg of PEG<sub>200</sub> at 40°C, then 50 mg of Solutol HS 15 was added (4.5:1:15,  
3 mPEG200/mSolutol HS 15/vPBS), and the mixture was gently homogenized for 30 min at 40°C. After  
4 this step, 750 µL of PBS at 40°C was added and the formula was gently mixed 10 min at  
5 40°C then 1h at room temperature ( $[(R)\text{-CE3F4}]_{\text{post process}} = 625 \mu\text{g/mL}$ ). In the case of lipid  
6 nanocapsules (LNC), a phase inversion process was used [47]. PBS was used as an aqueous  
7 phase, and a mixture of Solutol HS 15, EggPC, and Miglyol® as lipid phase. For 5 mg of (*R*)-  
8 CE3F4, the lipids proportions were 57.8% Mygliol, 37.0% Solutol HS 15 and 5.2% EggPC  
9 for a total amount of lipids of 2590 mg, and 1.1 mL of PBS. (*R*)-CE3F4 was first dissolved  
10 into Miglyol before the process. Both phases were mixed into a vial containing a magnet. The  
11 vial was put into a water bath, and three temperature cycles of heating/cooling between 55°C  
12 and 85 °C were applied. After reaching 85°C for the third time, the vial was cooled to 75°C  
13 and a dilution with 3 mL of cold PBS (5°C) was performed under magnetic stirring, leading to  
14 the spontaneous formation of lipid nanocapsules ( $[(R)\text{-CE3F4}]_{\text{post process}} = 900 \mu\text{g/mL}$ ).  
15 All three formulations were centrifuged at 4,000 g for 10 min to remove non-solubilized (*R*)-  
16 CE3F4. Supernatants were collected, (*R*)-CE3F4 concentration was measured and adjusted  
17 with PBS to obtain a final concentration of 625 µg/mL. The analytical method was applied to  
18 monitor the (*R*)-CE3F4 concentration in C57BL/6 plasma samples at 37°C.

### 19 **3 Results and discussion**

#### 20 *3.1 Spectral analysis of (R)-CE3F4 & method validation*

##### 21 *3.1.1 Spectral analysis*

22 Due to the lack of information on (*R*)-CE3F4, a preliminary study of the UV absorption  
23 spectra was performed in the range of 200 nm to 400 nm. This enables us to determine  
24 precisely the suitable wavelengths for HPLC-UV analysis. From the mean UV spectrum of  
25 (*R*)-CE3F4, two narrow bands, showing maxima at 202 nm and 220 nm, and two broad bands  
26 were observed (**Fig. S1**). The second derivative spectrum was calculated from the mean UV  
27 spectrum to exalt the spectrum signal and the two expected wavelengths from the broad bands  
28 were highlighted at 250 and 299 nm. Based on the variability of the measured absorbances  
29 (RSD %) at 202 nm, 220 nm, 250 nm, and 299 nm, which were 6.1 %, 2.12 %, 3.0 %, and  
30 10.4% respectively, we selected the wavelengths 220 nm and 250 nm for the quantification of  
31 (*R*)-CE3F4. Then both wavelengths were further investigated to improve the UV detection for

1 the quantitative method depending on the specificity of the chromatographic assay on human  
2 and murine plasma.

### 3 3.1.2 Validation of the (*R*)-CE3F4 method quantification in human plasma

4 The next step for conducting the CE3F4 bioanalysis was to set up a quantitative method using  
5 liquid chromatography. The relative hydrophobicity of CE3F4 (LogP= 3.6) allowed us to use  
6 a reversed stationary phase with a long alkyl chain (C18) and relative eluotropic strength  
7 (ACN/Water: 60/40 (v/v)). Accordingly, the method was validated in terms of specificity,  
8 linearity, LOQ, accuracy profile, and precision. Carryover, sample stability, and extraction  
9 recovery were also determined.

10 Under these analytical conditions, the (*R*)-CE3F4 had an elution time of 4.2 min and was  
11 well-separated from visible compounds from the human plasma matrix which were  
12 overwhelmingly eluted up to 3 min. Indeed, concerning the specificity, no signal was  
13 detectable at 220 or 250 nm on the 6 different blank human plasma samples close to CE3F4  
14 retention time (4.2 min) as displayed in **Fig. 2**. Similar results were obtained with enzyme  
15 inhibitors, or commonly used compounds for *in vivo* studies as isoflurane, Rompun®  
16 (xylazine), or the association of Zoletil® (tiletamine, zolazepam) at a lethal dose, or  
17 excipients for the formulation of the (*R*)-CE3F4 (PEG<sub>200</sub>, Solutol HS 15, EggPC, DSPE-  
18 PEG<sub>2000</sub>, Miglyol 812). At this stage, it is worth noting that two calibration curves  
19 corresponding to each selected wavelength (i.e. 220 and 250 nm) were performed. This  
20 strategy was applied to overcome the difference between species, with the possible  
21 interferences generated by matrix elements extracted from human or mouse plasma.

22 For the validation of the (*R*)-CE3F4 method quantification in human plasma, the  $\beta$ -  
23 expectation proportion of 80% for all QC level, using tolerance interval calculation, was more  
24 appropriate than that recommended by bioanalytical methods (67% of validated QC for at  
25 least 50% of QC levels fulfilling acceptance criteria). For QC, the validation domain ranged  
26 from 0.40 to 150  $\mu\text{g/mL}$  to cover the concentration range observed in the only *in vivo* study  
27 on CE3F4 [2]. However, these authors did not measure the concentration of CE3F4 in blood  
28 or plasma. An amount of 3 mg/kg was injected into C57BL/6 mice for an average total blood  
29 volume estimated to 72 mL/kg [48]. Therefore, the maximum blood concentration of CE3F4  
30 that could be observed should be 42  $\mu\text{g/mL}$  assuming a one-compartment model for blood  
31 (plasma + cells). However, if CE3F4 is only located in plasma (where the plasma volume is  
32 around 36-43 mL/kg for a usual hematocrit ranging from 40 to 50% [49]), the expected

1 maximum concentration should reach at least 83  $\mu\text{g/mL}$  for a dosage of 3 mg/kg.  
2 Consequently, a calibration curve covering concentrations from 0.40 to 150  $\mu\text{g/mL}$  was  
3 considered appropriate to evaluate the maximal concentration to 1% of the maximal expected  
4 concentration for the dosage of 3 mg/kg. In addition, such a calibration curve allows plasma  
5 quantification in plasma with the same sample treatment procedure for variable dosage  
6 regimens (i.e. no dilution step). These extended ranges led us to assess calibration curves  
7 from 0.40 to 200  $\mu\text{g/mL}$  with various weighted regressions including no weighing,  
8  $1/\text{concentration}$ , and  $1/\text{concentration}^2$ . The calibration curves were obtained without using  
9 internal standard since no appropriate CE3F4 analog was commercially available. Although  
10 each calibration curve provided an acceptable coefficient of determination  $R^2 (> 0.99)$ , only  
11  $1/\text{concentration}^2$  weighting fulfilled the relative error acceptance range for all levels (i.e.  
12 within  $\pm 20\%$  for the lowest concentration,  $\pm 15\%$  for other levels). The parameters of the  
13 calibration curves at 220 and 250 nm were determined (**Table 3**) and they satisfied the  
14 validation criteria, including their corresponding QCs (**Table 4**). Within-day and between-  
15 days accuracies and precisions were satisfactory.

16 Accuracy profiles were built from data combining precision and trueness (**Fig. S2**). At 220  
17 nm, biases were shown into the interval of  $\pm 10\%$ , and RSD  $< 10\%$  for LLOQ and  $< 5\%$  for  
18 other QCs. These values satisfied EMA criteria for the validation of bioanalytical methods.  
19 Based on these evaluated parameters, the corresponding accuracy profile at 220 nm has been  
20 built (**Fig. S2A**). The tolerance interval was between the limits given by the EMA ( $100 \pm 20\%$   
21 for LLOQ, and  $100 \pm 15\%$  for another QC level) for a  $\beta$ -expectation tolerance interval set at  
22 80%. These results allowed concluding that the method fulfilled the required reliability in  
23 terms of trueness and precision for the quantification of (*R*)-CE3F4 in the plasma matrix. The  
24 LLOQ and ULOQ were very satisfactory with values of 0.40  $\mu\text{g/mL}$  and 150  $\mu\text{g/mL}$ ,  
25 respectively. For the calibration curve at 250 nm, very similar results about biases and RSD  
26 were obtained (biases are into the interval of  $\pm 10\%$ , and RSD  $< 10\%$ ). The corresponding  
27 accuracy profile is reported in **Fig. S2B**, where it appears that the tolerance interval for LLOQ  
28 was near the upper acceptance limit but was still conforming (119.8%). Finally, the  
29 calibration curves for human plasma at the two wavelengths satisfied the EMA criteria. Also,  
30 the method demonstrated a strong ability to accurately quantify (*R*)-CE3F4, even considering  
31 the tolerance interval for each QC level. For the carryover, the signal was less than 20% of the

1 LLOQ on the first injection of blank plasma at the retention time of the analyte (no peak  
2 detectable) after the injection of the highest concentration used for calibration (200 µg/mL).  
3 Besides, the extraction of CE3F4 was assessed from calibration curve “A” (i.e. CE3F4  
4 addition into plasma before the extraction process ) and calibration curve “B” as reference  
5 (i.e. CE3F4 addition after extraction process on blank plasma). The simple extraction  
6 procedure, which consisted to precipitate proteins with 150 µL of ACN, followed by a  
7 centrifugation step, allowed a mean extraction recovery on three days of 100.4% with an RSD  
8 of 6.4%. Consequently, the use of an internal standard was unnecessary due to a total  
9 extraction, a low RSD, and UV detection, contrarily to mass spectrometry, which involves an  
10 ionization yield in the case of a coelution with molecules from the complex matrix, justifying  
11 the use of an internal standard to reduce variability and correct bias (i.e. isotope-labeled  
12 CE3F4).

### 13 *3.1.3 Human plasma samples stability*

14 The stability of the extracted samples was investigated from the validated calibration curve.  
15 After 15 h in an autosampler, the bias between the first and second analysis session was  
16 superior to 15 % for samples that were first injected due to solvent evaporation, but the  
17 stability was correct for non-immediately injected samples (less than 15% loss). Concerning  
18 stock solutions stored at 5°C for one year, QCs stored at -20°C for one month, or exposed to 3  
19 cycles of freeze-thaw, or after two hours at room temperature, degradation was not significant  
20 (< 5% for stock solutions, < 15% for all conditions using plasma). It can be concluded that  
21 such results make it possible to obtain sufficient stability to organize analyses by a campaign,  
22 at a still experimental stage of pharmaceutical development. More long-term stability could  
23 be explored if the (*R*)-CE3F4 exhibits a clinical interest to be tested in humans.

### 24 *3.2 Application of the LC-UV method to the evaluation of (R)-CE3F4 ex vivo stability in* 25 *human or murine plasma*

26 Various risk points were verified to check the transposability of the HPLC-UV method, in  
27 particular concerning the specificity of the signal, and the extraction efficiency of CE3F4.

#### 28 *3.2.1 Chromatograph difference in specificity between human and murine plasma*

29 As previously mentioned, the results on specificity showed no detectable signal near CE3F4  
30 at 220 or 250 nm on blank human plasma samples, but differences resulted with blank murine  
31 plasma samples. Indeed, visible peaks at the CE3F4 retention time (4.2 min) were observed in  
32 these matrices at 220 nm (**Fig. 3A**). When switching to 250 nm, detection specificity was



1 enhanced and thus satisfactory, without any visible coelution impacting CE3F4 quantification  
2 (**Fig. 3B**). Different gradient conditions were studied (results not shown), but none of them  
3 adequately improved the resolution at 220 nm in addition to extending the time analysis (until  
4 one hour). These results are in agreement with our previous hypothesis on the interest to  
5 validate several detection conditions when UV detection is used (i.e. 220 and 250 nm)  
6 because of the possible interferences on chromatograms depending on the origin of the  
7 plasma (i.e. mouse or human). Therefore, the wavelength of 250 nm should be used for  
8 quantification in mouse plasma, and both 220 and 250 nm are suitable for human plasma  
9 analysis.

### 10 3.2.2 (*R*)-CE3F4 and *ex vivo* plasmatic metabolism at 37°C

11 We first investigated *ex vivo* the (*R*)-CE3F4 behavior in the plasmas from two species as a  
12 function of the time to evaluate a potential plasma metabolism at 37°C. To be suitable for  
13 HPLC analysis, samples were treated as described previously, and chromatograms were  
14 collected using wavelengths 220 and 250 nm for human and mouse plasma, respectively. The  
15 chromatographic analysis of the murine and human plasma samples showed a peak at 8.6 min  
16 (TRR = 2.05) at 220 nm and 250 nm (**Fig. 4**) in addition to CE3F4 (4.2 min). Also, a lack of  
17 specificity was observed at 220 nm at 4.2 min in murine plasma (**Fig. 4A**), by using the  
18 appropriate wavelength detection at 250 nm, the peak was recovered, confirming the  
19 usefulness of this wavelength for preclinical studies in mice.

20 A significant difference in the kinetics of the disappearance of the main peak was observed  
21 between the two plasma samples. Indeed, a decrease of 23 % of the peak area was observed in  
22 human plasma after 24 h (**Fig. 5A**). In contrast, a decrease of 95 % was found in murine  
23 plasma after only 3 h, and no peak at the expected time was detected after six hours  
24 confirming the total degradation of CE3F4 (**Fig. 5A**). These results suggest that (*R*)-CE3F4  
25 undergoes distinct kinetic degradation. The observed half-life in murine plasma was 39.9 min  
26 (confidence interval 95%: from 36.9 to 43.4 min) against a half-life superior to 24 h in human  
27 plasma. As depicted in **Fig. 5B**, the CE3F4 related metabolite showed two phases in murine  
28 plasma, an acute ascending phase, and a slow decreasing phase suggesting the instability of  
29 the metabolite in murine plasma (e.g. due to the formation of secondary metabolites, or non-  
30 enzymatic degradation such as oxidation reaction). For human plasma, this metabolite  
31 appeared very slowly.

1 We therefore hypothesized that enzyme-dependent hydrolysis promoted metabolism on the  
2 (*R*)-CE3F4 N-formyl group (**Fig. S3**) resulting in a more apolar compound than (*R*)-CE3F4  
3 which is coherent with the HPLC analysis exhibiting a longer elution time than (*R*)-CE3F4 on  
4 the reverse-phase column. The spectral analysis of this degraded CE3F4 showed similar  
5 features with the original CE3F4 spectrum depicting three maxima (218 nm, 250 nm, and 325  
6 nm), indicating a structural similarity with the CE3F4 (**Fig. S4**). Among the known CE3F4  
7 derivatives, deformed-CE3F4 may be a candidate. To confirm this structural hypothesis,  
8 LC-UV and LC-MS/MS analyses were performed on synthesized deformed-CE3F4 as  
9 described in a specific part below (3.3). It is worth mentioning that the deformed-CE3F4  
10 was inactive on EPAC1 [3]. On this metabolism through N-deformylation, Gleason *et al.*  
11 reported an inter-species variability on the plasmatic metabolism of diformyldapsone.  
12 Different kinetics of metabolism have been described in plasmas, in the following order:  
13 mouse > rat > guinea pig > rabbit, while no N-deformylation in the dog or human plasma was  
14 observed [19]. Some authors observed differences in plasma metabolism between rodents,  
15 dogs, and humans for an N-formylated indolocarbazole topoisomerase I inhibitor [20], with  
16 no N-deformylation in dogs or humans plasma. Other studies also reported similar differences  
17 between species in the type of plasmatic esterases [23,38,39,50]. They noticed the absence of  
18 carboxylesterases in human plasma in contrast to rodent plasmas. However, liver metabolism  
19 showed an inversed tendency for humans and rodents. Another study on diformyldapsone  
20 with liver homogenates changed the rank order of metabolism rates: guinea pig  $\geq$  human >  
21 mouse = rabbit = rat > dog [18]. Based on these results, it appears evident to consider enzyme  
22 inhibitors to stabilize samples following plasma collection, for *in vivo* kinetic experiments,  
23 especially in mice.

### 24 3.2.3 Stabilization studies of (*R*)-CE3F4 in murine plasma at 37°C with paraoxon

25 To proceed to the (*R*)-CE3F4 quantification in murine plasma, different strategies were  
26 explored using either EDTA, paraoxon, BNPP, or benzil. EDTA can inhibit metal-dependent  
27 enzymes, such as hydrolases including e.g. amidohydrolase [51] or paraoxonase [23,52,53] in  
28 addition to being of interest as an anticoagulant for sample collection. Paraoxon is known as a  
29 well-known powerful esterase inhibitor (organophosphate oxon), as described by Gleason and  
30 *al.* [19] for diformyldapsone stabilization in plasma samples. BNPP and benzil are more  
31 specific inhibitors than paraoxon. BNPP is a specific inhibitor of carboxylesterases [54,55],

1 phosphodiesterase [56]. Benzil is a specific reversible inhibitor of mammalian  
2 carboxylesterases [57].

3 EDTA did not exhibit any stabilizing effect against CE3F4 degradation in murine plasma  
4 samples. Thus, the enzyme involved in CE3F4 degradation is not metal-dependent.  
5 Nevertheless, it could be interesting to note that EDTA can inhibit the paraoxonase at 5 mM  
6 [23] and be used in this study at 10 mM. The enzyme is involved in the metabolism of  
7 organophosphate like paraoxon.

8 Paraoxon was tested at different concentrations to stop the degradation of CE3F4. It has been  
9 shown that plasmatic N-deformylation was inhibited with paraoxon. Firstly, paraoxon was  
10 added in the plasma 5 min before the (*R*)-CE3F4 for homogenization of the sample to allow at  
11 least a 1-hour blockade. This 1-hour delay was chosen to mimic conditions of sample  
12 treatment (i.e. blood sampling, centrifugation step for plasma/serum separation, until protein  
13 precipitation by solvents). The inhibition of CE3F4 degradation was null, partial, and total for  
14 paraoxon at  $10^{-6}$  M,  $10^{-5}$  M, and  $10^{-4}$  M, respectively (**Fig. 6**). These results confirmed the  
15 relevance of paraoxon to inhibit plasmatic deformylation of CE3F4 and subsequently, the  
16 stabilization of plasma samples. Moreover, the paraoxon experiments suggest the involvement  
17 of a hydrolase in CE3F4 metabolism. Secondly, when  $10^{-4}$  M paraoxon was added  
18 simultaneously with (*R*)-CE3F4 to mimic real sample treatment (i.e. addition of paraoxon as  
19 first pre-treatment step after blood sampling), no CE3F4 degradation was observed and no  
20 metabolite was detected on the chromatograms (data not shown).

21 BNPP and benzil were assayed, but they did not stabilize the CE3F4 against degradation on a  
22 1-hour duration (**Table 5**). A slight concentration-dependent inhibition was observed with  
23 BNPP, but it appears that higher concentrations should be considered to obtain sufficient  
24 inhibition, which is of no interest in practice. These results exhibit that carboxylesterases are  
25 not involved in CE3F4 degradation. Consequently, the paraoxon at the inhibiting  
26 concentration of  $10^{-4}$  M was retained to allow mouse sample pre-treatment.

#### 27 *3.2.4 Validation of the (R)-CE3F4 method quantification in murine plasma loaded with* 28 *paraoxon*

29 Based on the validation procedure in human plasma, and the stabilization of CE3F4 in murine  
30 plasma thanks to paraoxon, the validation of the (*R*)-CE3F4 quantification method at 250 nm  
31 in murine plasma loaded with paraoxon has been carried out. As for validation in human  
32 plasma, only  $1/\text{concentration}^2$  weighting fulfilled the relative error acceptance range for all

1 levels (i.e. within  $\pm 20\%$  for the lowest concentration,  $\pm 15\%$  for other levels). The parameters  
2 of the calibration curves at 250 nm were determined (**Table 6**) and shown to satisfy the  
3 validation criteria.

4 As observed in human plasma, no carryover was observed on the first injection of blank  
5 murine plasma at the retention time of the CE3F4 (no peak detectable) after the injection of  
6 the highest concentration used for calibration (200  $\mu\text{g}/\text{mL}$ ), and the mean extraction recovery  
7 on three days was 100.5% with an RSD of 2.3%. Then paraoxon did not influence the CE3F4  
8 extraction in murine plasma. Because of similarities of results between human plasma and  
9 murine plasma loaded with paraoxon, three QC levels were used to rationalize the use of  
10 animals for the transposition of the already fully validated method in human plasma. The  
11 results on QCs were satisfactory (**Table 7**) because all biases were shown in the interval of  $\pm$   
12 5%, and RSD  $< 5\%$ . The corresponding accuracy profile is reported in **Fig. S7**.

13 Thus, the validated method using human plasma allowed to manage murine samples, and then  
14 the method was validated on three days in murine plasma loaded with paraoxon allowing the  
15 quantification of CE3F4 at 250 nm from 0.40 to 150  $\mu\text{g}/\text{mL}$ .

#### 16 3.2.5 *Murine sample storage*

17 For the stability of murine plasma spiked with paraoxon  $10^{-4}$  M, a critical degradation of the  
18 (*R*)-CE3F4 was observed after three days at  $5^{\circ}\text{C}$  (-79.6%). However, no significant variations  
19 ( $< 15\%$ ) were observed at  $-20^{\circ}\text{C}$  after 1 week, after 3 cycles of freeze-thaw, after two hours at  
20 room temperature, or in the plasma extracts after 15h in the autosampler. This could be  
21 explained by the presence of paraoxonase in murine plasma, which degrades paraoxon and  
22 then allows the hydrolysis of CE3F4. This hypothesis could be surprising because paraoxon is  
23 known as an irreversible inhibitor. However, a spontaneous reactivation of mouse  
24 butyrylcholinesterase after paraoxon inhibition has been described by Li et al. [23], and pieces  
25 of evidence of other inhibition mechanisms such as the allosteric modification or steric  
26 hindrance which can be reversible were reported [58–60].

27 Thus, storage at  $-20^{\circ}\text{C}$  facilitates the practical aspect of handling biological samples, allowing  
28 flexibility in analysis schedules. Storage at  $-80^{\circ}\text{C}$  should be considered and evaluated for the  
29 long-term storage of biological samples.

### 30 3.3 *Identity of the main metabolite*

#### 31 3.3.1 *HPLC-UV analysis*

1 In addition to the efficacy of paraoxon in inhibiting the degradation of CE3F4 in murine  
2 plasma, similarities were observed between unknown metabolites and deformed-CE3F4.  
3 Samples were prepared to obtain relatively close concentration (i.e. deformed-CE3F4 and  
4 (*R*)-CE3F4 at 50 µg/mL in ACN versus 100 µg/mL of (*R*)-CE3F4 in murine plasma analysed  
5 after a little more than a half-life). The RP-LC analysis resulted in similar relative retention  
6 times: 2.05 and 2.08 for deformed-CE3F4 and observed metabolite, respectively (**Fig.**  
7 **S5A**). A small peak corresponding to a synthesis impurity of (*R*)-CE3F4 was detected at 5.2  
8 min. When analysing the UV spectra of deformed-CE3F4 and metabolite, the high  
9 similarity was obtained with three identical maxima at 218 nm, 250 nm, and 322 nm (**Fig.**  
10 **S5B**). These results supported the hypothesis that the metabolite observed in murine plasma  
11 was the deformed-CE3F4.

### 12 3.3.2 HPLC-MS/MS analysis

13 To undoubtedly confirm the structure of CE3F4 metabolite as deformed-CE3F4 and so the  
14 mechanism involved in CE3F4 degradation, mass spectrometry (MS) experiments were  
15 performed. Deformed-CE3F4 presents a molar mass of 322.9 g/mol which corresponds to  
16 a parent ion of *m/z* of 323.9 in positive mode [*MH*<sup>+</sup>]. Total ion chromatograms (TIC) and  
17 extracted chromatograms (*m/z* 323.9, *m/z* 351.9) of plasma sample, deformed-CE3F4 and  
18 (*R*)-CE3F4 controls are presented in **Fig. 7**. It should be noted that because the flow rate of  
19 the mobile phase is 0.5 mL/min instead of 1.0 mL/min in previous experiments, the time of  
20 retention of analytes was doubled. Indeed, (*R*)-CE3F4 is eluted at 8.4 minutes, whereas the  
21 unknown metabolite is eluted at 18.3 minutes. The unknown metabolite and deformed-  
22 CE3F4 are eluted with the same retention time. Regarding MS experiments, the hypothesized  
23 metabolite and deformed-CE3F4 exhibited the same parent ion of *m/z* 323.9 as observed  
24 on the chromatogram of the extracted ion *m/z* 323.9 (**Fig. 7A & C**) and the comparison of the  
25 MS spectrum extracted at 18.3 min of plasma sample (**Fig. 8A**) and deformed-CE3F4  
26 (**Fig. 8B**). Also, the isotopic distribution is similar and is coherent with the presence of two  
27 bromine (**Fig. 8C & D**). Finally, both the unknown metabolite and deformed-CE3F4  
28 showed the same MS fragmentation pattern of the parent ion (**Fig. 8E & F**). Altogether, these  
29 results confirmed that the metabolite observed in murine plasma was the deformed-  
30 CE3F4. MS and MS/MS spectrum of CE3F4 were also recorded and are presented in  
31 supplemental data (**Fig. S6**).

1 3.4 *Application to the evaluation of (R)-CE3F4 protection against enzymatic degradation*  
2 *by various formulations.*

3 For all formulations, the degradation of (R)-CE3F4 in plasma followed an exponential decay  
4 ( $R^2 > 0.99$ ), allowing the estimation of an apparent half-life linked to the ability of the  
5 formulation to protect the (R)-CE3F4 against enzymatic degradation. As synthesized in **Table**  
6 **8**, the formulations exhibited different protection levels. When solubilized in micelles, the  
7 (R)-CE3F4 degradation was equivalent to (R)-CE3F4 solubilized in DMSO (apparent half-life  
8 of 0.8 h), indicating a lack of protection of the (R)-CE3F4 by this formulation. The Solutol  
9 HS 15/PEG<sub>200</sub> mixture showed an intermediate level of protection, with a (R)-CE3F4 apparent  
10 half-life of 2.7 h. Lipid nanocapsules provided the most important protection effect with an  
11 apparent half-life of 5.5 h for (R)-CE3F4. Thus, the analytical method described in this study  
12 proved efficient for the identification of promising formulations able to optimize (R)-CE3F4  
13 administration for future pharmaceutical development.

14 **4 Conclusion**

15 The quantification method including a simple extraction protocol showed to be well suitable  
16 for the quantification of the (R)-CE3F4 in plasma for preclinical studies as well as clinical  
17 studies. This method allows drug quantification in complex matrices from different origins  
18 providing that appropriate wavelength selection is performed. Because of the impact of the  
19 matrix observed at 220 nm in murine plasma that is still unresolved despite the application of  
20 different gradient conditions, an adaptation of the wavelength in agreement to the studied  
21 species was necessary (220 nm or 250 nm for human samples, only 250 nm for mouse  
22 samples) highlighting the importance of the wavelength selection for the (R)-CE3F4  
23 quantification.

24 The validated method allows the processing of the samples and the rendering of the results  
25 within the same working day. Moreover, it permits to follow the main metabolite of (R)-  
26 CE3F4 observed in mouse plasma within the same analysis. The metabolite was identified as  
27 the N-deformylated-CE3F4 and the involvement of a hydrolase was revealed by paraoxon use  
28 and further by HPLC-UV and HPLC-MS/MS studies.

29 We also show the interest of paraoxon in preclinical studies to stop metabolic degradation  
30 during sample pretreatment in mouse models. Future studies are needed to identify the  
31 enzyme responsible for the observed degradation, for the possible development of knock-out  
32 mice models lacking this enzyme [61] for pharmacokinetic and efficacy studies.

1 Furthermore, non-plasma-based metabolic pathways should be considered, like metabolic  
2 stability assay on microsomes or hepatocytes knowing that hepatic metabolism is central in  
3 humans, to elucidate the (*R*)-CE3F4's metabolic profile and then identify any novel  
4 metabolites and metabolic pathways in humans and mice. To conclude in an objective of  
5 pharmaceutical development, it appears essential to provide for the protection of the (*R*-  
6 CE3F4 against all potential enzymatic degradations (i.e. also in the liver for humans), and to  
7 develop an appropriate formulation accordingly.

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11 Therapeutics (ANR-10-LABX-0033-LERMIT).

## 12 **Conflict of Interest**

13 The authors declare no conflict of interest regarding this publication.

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6

7

1 Table 1.

2 Concentrations ( $\mu\text{g/mL}$ ) for the eight calibration standards, the six quality controls in human  
3 plasma, and the three quality controls in mouse plasma.

Sample type	Number of levels	[CE3F4] for each level ( $\mu\text{g/mL}$ )							
Calibration	8	0.40	0.97	2.36	5.74	13.9	33.9	82.3	200
Human quality controls	6	0.40	1.50	5.00	25.0	60.0	150		
Murine quality controls	3	0.40	8.00	150					

4

5 Table 2.

6 Detail of volumes ( $\mu\text{L}$ ) used for the sample preparation, for each working solution  
7 (calibration standard, CS; and human or murine quality control, QC), plasma (human or  
8 murine), and solvents.

Sample type	CS	QC	Plasma with unknown [(R)-CE3F4]	Blank plasma	EtOH	ACN for extraction
Calibration	50	-	-	50	-	150
QC	-	50	-	-	50	150
Plasma samples (human or murine)	-	-	50	-	50	150

9

10 Table 3.

11 Average parameters of the linear regression obtained from three calibration curves of the  
12 analytes with a  $1/\text{concentration}^2$  weighting at 220 or 250 nm.

Wavelength	n	Linear range	$R^2$ ( $\pm\text{SD}$ )	Slope ( $\pm\text{SD}$ )	y-Intercept ( $\pm\text{SD}$ )
220 nm	3	0.40-200 $\mu\text{g/mL}$	0.9967 ( $\pm 0.0008$ )	14585.5 ( $\pm 773.1$ )	713.1 ( $\pm 987.6$ )
250 nm	3	0.40-200 $\mu\text{g/mL}$	0.9972 ( $\pm 0.0025$ )	4136.7 ( $\pm 113.7$ )	-174.1 ( $\pm 143.6$ )

13

14

1 Table 4.

2 Within -day and between days accuracies and precisions on six quality control for  
3 quantification at 220 nm (A) and 250 nm (B). CV: coefficient of variation.

(A)	QC level	QC1	QC2	QC3	QC4	QC5	QC6
Within-day accuracy (%), n = 5 per day	Day 1	7.35	-1.46	-4.56	-8.12	0.49	5.11
	Day 2	4.17	-6.98	-7.76	-12.22	-5.13	-2.31
	Day 3	2.10	1.67	-2.56	-6.29	1.52	2.41
Within-day CV (%)		7.95	3.20	3.02	2.26	2.55	2.44
Between-days accuracy (%), n = 3		17.00	-2.94	-9.21	-11.88	-5.01	1.93
Between-days CV (%)		5.01	10.38	3.08	2.99	2.54	6.64
(B)	QC level	QC1	QC2	QC3	QC4	QC5	QC6
Within-day accuracy (%), n = 5 per day	Day 1	5.18	1.28	-4.49	-9.69	-1.43	2.94
	Day 2	9.84	-6.86	-6.39	-8.32	-0.50	2.65
	Day 3	4.80	1.88	-2.46	-5.89	3.10	4.19
Within-day CV (%)		8.86	3.72	3.06	2.18	2.42	2.26
Between-days accuracy (%), n = 3		19.20	2.74	-7.42	-11.03	-1.07	5.15
Between-days CV (%)		6.01	9.41	3.44	2.61	5.62	0.72

4

5 Table 5.

6 Stability of (R)-CE3F4 in mouse plasma sample in presence of enzyme inhibitors BNPP or  
7 benzil.

Time (h)	[BNPP] (M)			[Benzil] (M)		
	$10^{-4}$	$10^{-5}$	$10^{-6}$	$10^{-4}$	$10^{-5}$	$10^{-6}$
0	100%	100,00%	100,00%	100,00%	100,00%	100,00%
0,5	68,2%	67,7%	56,1%	56,8%	60,4%	59,4%
1	56,6%	41,8%	37,6%	33,7%	35,2%	36,4%

8

9

1 Table 6.

2 Average parameters of the linear regression obtained from three calibration curves of the  
3 analytes with a  $1/\text{concentration}^2$  weighting at 220 or 250 nm.

Wavelength	n	Linear range	R <sup>2</sup> (±SD)	Slope (±SD)	y-Intercept (±SD)
250 nm	3	0.40-200 µg/mL	0.9964 (±0.0015)	5339.8 (±90.0)	-93.0 (±184.1)

4

5 Table 7.

6 Within-day and between days accuracies and precisions on three quality control for  
7 quantification at 250 nm in murine plasma. CV: coefficient of variation.

	QC level	QC1	QC2	QC3
Within-day accuracy (%), n = 5 per day	Day 1	-3.93	-6.40	-7.40
	Day 2	0.78	-7.70	-9.71
	Day 3	-2.14	-2.13	-7.80
Within-day CV (%)		4.30	4.37	1.08
Between-days accuracy (%), n = 3		-4.11	-2.68	-9.65
Between-days CV (%)		7.68	10.23	1.82

8

9 Table 8.

10 Apparent half-life in murine plasma at 37°C of the (*R*)-CE3F4 in different formulations.

Formulation	Apparent half-life (h)
DSPE-PEG <sub>2000</sub> micelles	0.8
Solutol HS 15 & PEG <sub>200</sub> mixture	2.7
Lipid nanocapsules	5.5

11

1 **Figure caption**

2 Figure 1.

3 Structure of (*R*)-CE3F4.

4

5 Figure 2.

6 Chromatograms of blank human plasma extract at 220 nm (A) and 250 nm (B).

7

8 Figure 3.

9 Chromatograms of blank mouse plasma extract at 220 nm (A) and 250 nm (B)

10

11 Figure 4.

12 Chromatograms at 220 nm (A) and 250 nm (A') of (*R*)-CE3F4 (TR = 4.2 min) and its  
13 principal metabolite (TR = 8.6 min) in murine plasma, after 3h at 37°C. Chromatograms at  
14 220 nm (B) and 250 nm (B') in human plasma, after 24h at 37°C.

15

16 Figure 5.

17 (A) Kinetics of the disappearance of the (*R*)-CE3F4 based on signal variation in human (at  
18 220 nm) or murine (at 250 nm) plasma. Continuous lines represent the mean regression and  
19 the dotted lines represent the confidence interval on the regression (95%). (B) Kinetics of the  
20 principal metabolite of the (*R*)-CE3F4 (TR = 8.6 min) observed at 250 nm in murine and  
21 human plasma at 37°C.

22

23 Figure 6.

24 Effect of paraoxon on (*R*)-CE3F4 metabolism in murine plasma at 37°C. Paraoxon was at  
25 different concentrations added 10 minutes before the addition of (*R*)-CE3F4 (blue, orange,  
26 and red lines), or simultaneously (green line).

27

28 Figure 7.

29 Chromatograms of (A) murine plasma sample containing (*R*)-CE3F4 and metabolite, (B) (*R*)-  
30 CE3F4 at 5 µg/mL, and (C) deformedylated-CE3F4 at 5 µg/mL. Total ion chromatogram is  
31 presented as a continuous black line, extracted ion chromatogram of m/z 351.9 is presented as



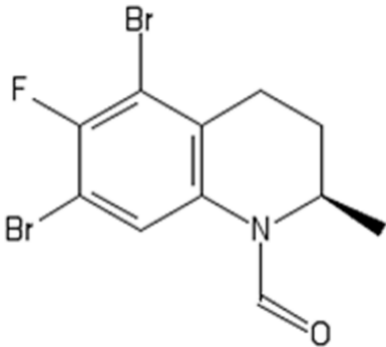
1 a dotted red line, and extracted ion chromatogram of  $m/z$  323.9 is presented as a dotted blue  
2 line.

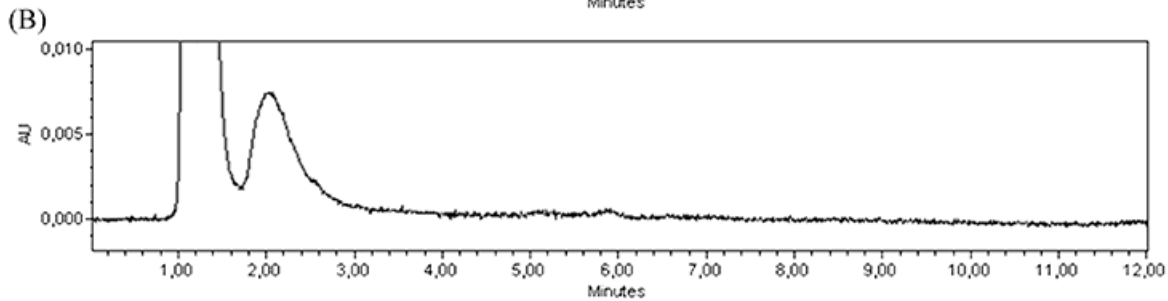
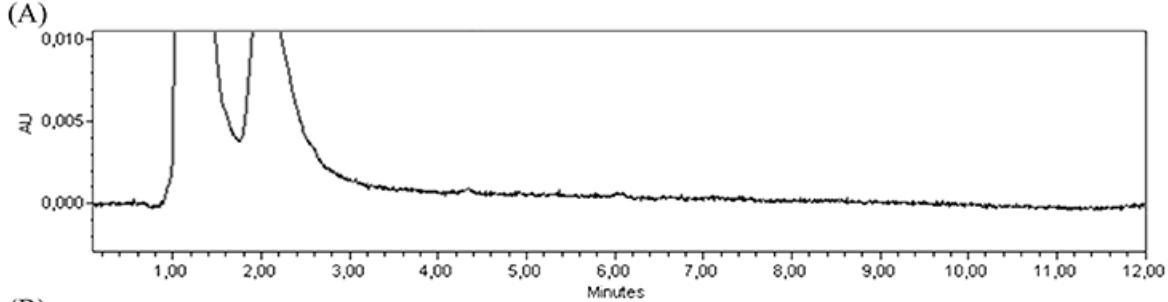
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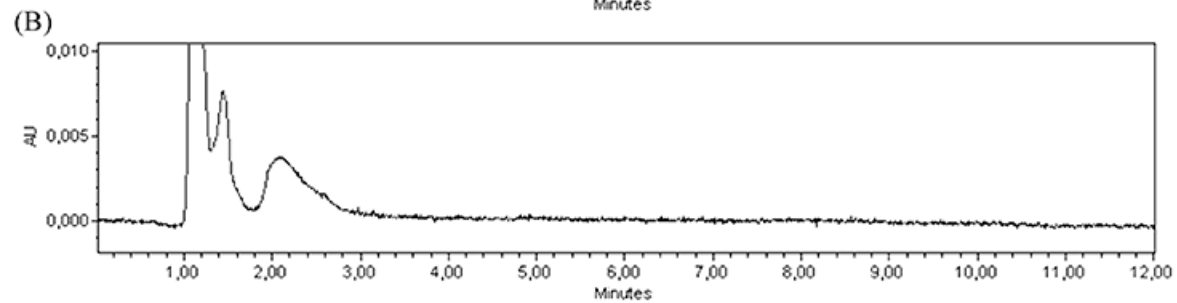
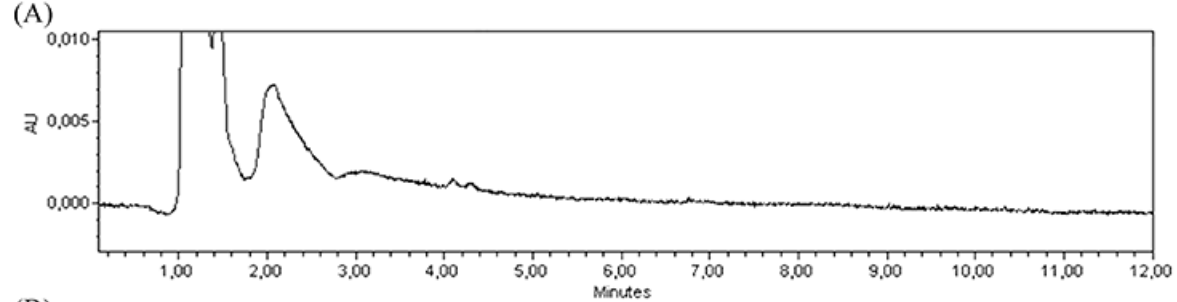
4 Figure 8.

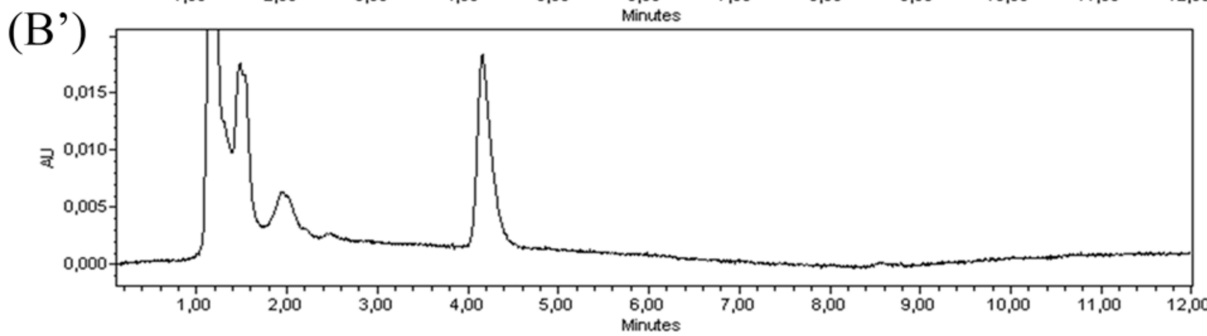
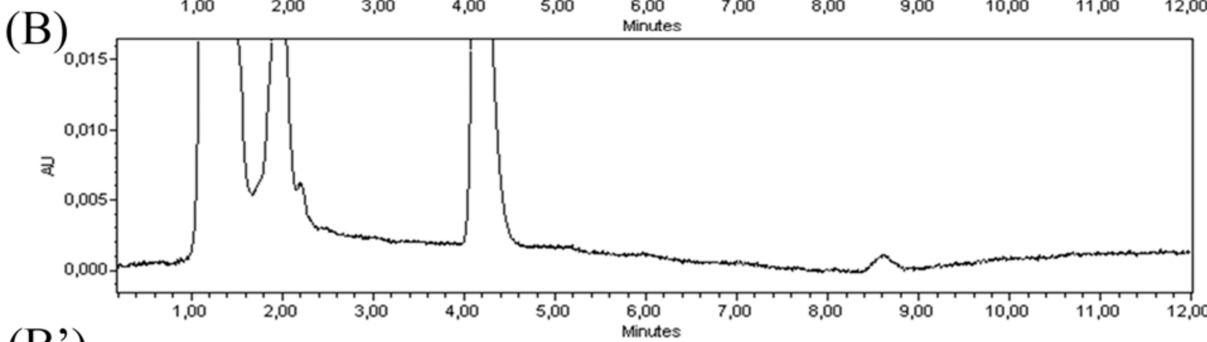
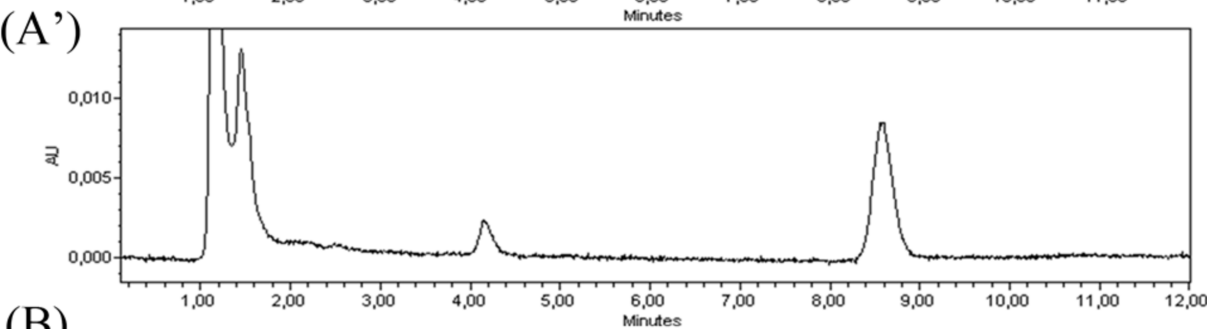
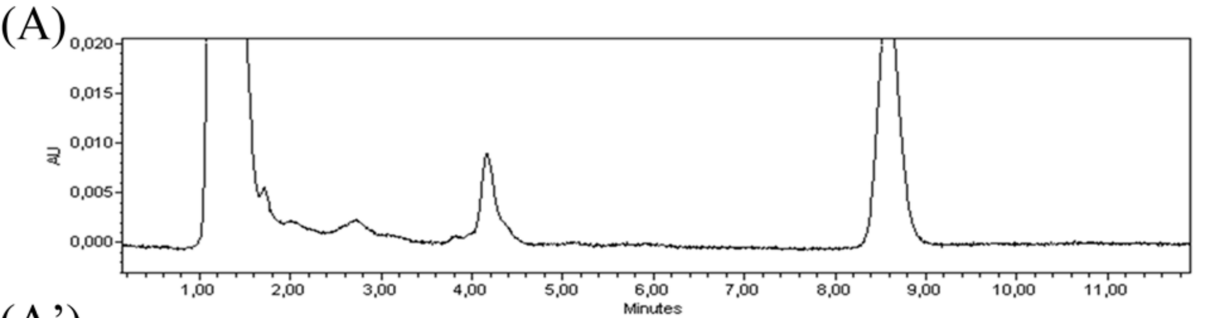
5 MS spectrum extracted at 18.3 min, isotopic distribution of  $m/z$  323.9 and MS/MS spectrum  
6 of the parent ion  $m/z$  323.9, in plasma sample (A, C and E respectively) and a control sample  
7 of deformylated-CE3F4 (B, D, and F respectively).

8

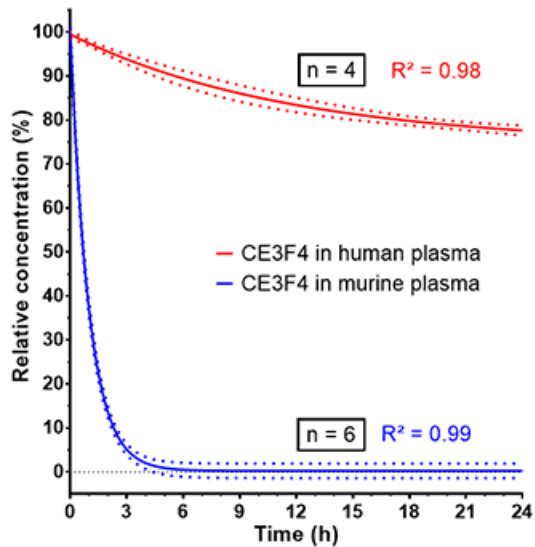








(A)



(B)

