Two or three things I am sure about uncertainty

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TUMLA, Rennes (2022-07-06)



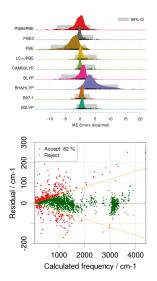
- Oncertainty validation
- Statistical treatment of Model Inadequacy

4 Conclusions

whoami

- Permanent CNRS researcher (orcid:0000-0001-8586-6222) (https://tinyurl.com/52zbchaz)
- Main research topics
 - Statistics for benchmarking of comput. chemistry methods
 - Uncertainty management in complex chemical models
 - Uncertainty quantification in computational chemistry





Preliminaries

Uncertainty vs. probability



Representation of uncertainty

- one can attribute a "degree of confidence" p(X = x)to any possible value x of an uncertain quantity X^1
- consistency and normalization $\Rightarrow p(X = x)$ follows the laws of probability²

• This approach at the core of the GUM $(3.3.1)^3$

- "measurement uncertainty reflects the lack of knowledge about the true value of a measurand"
- "the corresponding state-of-knowledge is best described by a distribution on the possible values of the measurand"
- The Maximum Entropy Principle (Maxent) states that, among all distributions compatible with a state-of-knowledge, on should select the one which minimizes Shannon entropy H(p).

¹P-S. Laplace (1825) Essai philosophique sur les probabilités, 5th ed. ²Dupré, Maurice J., Tipler, Frank J. (2009) Bayesian Analysis 3:599-606 ³JCGM 100:2008 Evaluation of measurement data – Guide to the expression of uncertainty in measurement. (ISO/IEC Guide 98-3)

Uncertainty vs error⁴



In order to estimate a measurement uncertainty it is assumed that the result of a measurement has been corrected for all recognized significant systematic effects and that every effort has been made to identify such effects.

systematic error

component of measurement error that remains constant or varies in a **predictable** manner

random error

component of measurement error that varies in an **unpredictable** manner

uncertainty

non-negative parameter characterizing the **dispersion** of the quantity values being attributed to a measurand

• often estimated by a standard deviation u, or the half-width of a probability interval U_p

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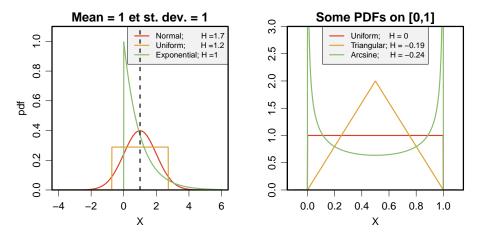
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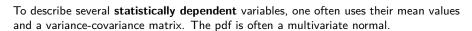
⁴Guide to the expression of uncertainty in measurement (GUM), JCGM **100**:2008, International Vocabulary of Metrology (VIM), JCGM **200**:2012

Preliminaries



Examples: entropy vs. pdf





$$p(x, y | \mu_{X_1}, \mu_{X_2}, \boldsymbol{\Sigma}) \propto (\det \boldsymbol{\Sigma})^{-1/2} \exp\left(-\frac{1}{2} \boldsymbol{E}^T \boldsymbol{\Sigma}^{-1} \boldsymbol{E}\right)$$

$$\boldsymbol{E} = \begin{pmatrix} x_1 - \mu_{X_1} \\ x_2 - \mu_{X_2} \end{pmatrix}; \boldsymbol{\Sigma} = \begin{pmatrix} u_{X_1}^2 & u_{X_1X_2}^2 \\ u_{X_2X_1}^2 & u_{X_2}^2 \end{pmatrix} = \boldsymbol{S}^T \boldsymbol{R} \boldsymbol{S}$$
$$\boldsymbol{S} = \begin{pmatrix} u_{X_1} \\ u_{X_2} \end{pmatrix}; \boldsymbol{R} = \begin{pmatrix} 1 & \rho \\ \rho & 1 \end{pmatrix}$$

with

- Σ is the variance-covariance matrix
- **R** is the correlation matrix $(-1 \le \rho \le 1)$



Sum-constraints



The case of conservation laws

• N variables such as $\sum_{i=1}^N X_i = 1$ et $\{X_i \ge 0; \, i=1,N\}$, one uses $Dirichlet's \ law$ (Maxent)

 $\{X_1,\ldots,X_N\} \sim Diri(\gamma * (\mu_1,\ldots,\mu_N))$

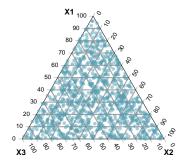
where $\gamma > 0$ is a precision factor (variance increases as γ decreases)

• examples: chemical compositions, branching ratios⁵...

⁵Plessis et al. (2010) J. Chem. Phys. 133:134110

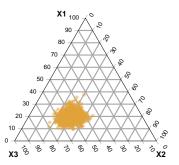
Dirichlet pdf

Diri(3 * (1/3,1/3,1/3))





Diri(100 * (0.2,0.3,0.5))





Quantifying uncertainty on rate constants

chemical equations based on empirical parameters

 $A + B \longrightarrow C + D; k_{AB}(T, P, M)$

$$\frac{da(t)}{dt} = -k_{AB}(T, P, M) a(t) b(t)$$

- reaction rates from experiments
 - \rightarrow with [[very] large] uncertainty
 - \rightarrow numerous estimated parameters

 \rightarrow in Titan's chemistry models, less than 10% of rate constants have been measured in representative (T,P,M) conditions...

\rightarrow modelers need extrapolation (with uncertainty !)

Arrhenius rate analysis (for extrapolation)



Data

$$\{T_i, k_i, u_{k_i}\}_{i=1}^N$$

Linear regression

$$\ln k_i = \ln k_0 - E/T_i$$
$$y = a + bx$$

with x=-1/T and uncertainties $u_{y_i}=u_{k_i}/k_i$ corresponding to the log-transformation of the rate constants

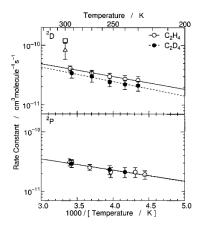
Uncertainty propagation

$$u_{\ln k}(T) = \left(u_a^2 - \frac{2}{T}u_a u_b \rho_{ab} + \frac{1}{T^2}u_b^2\right)^{1/2}$$
$$F_k(T) = \exp\left(u_{\ln k}(T)\right)$$

Example: $N(^{2}D) + C_{2}H_{4}$



Sato et al. (1999) Phys. Chem. A 103:8650-8656



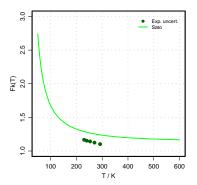
A/cm ³ molecule ⁻¹ s ⁻¹	$E_a/kcal mol^{-1}$	ref
$(2.3 \pm 0.3) \times 10^{-10}$	1.0 ± 0.1	this work
$(2.4 \pm 0.5) \times 10^{-10}$	1.1 ± 0.1	this work
$(7.1 \pm 4.4) \times 10^{-11}$	1.5 ± 0.3	32
$(3.3 \pm 1.4) \times 10^{-11}$	1.4 ± 0.1	32
$(1.6 \pm 0.2) \times 10^{-10}$	0.53 ± 0.06	33
$(1.4 \pm 0.1) \times 10^{-10}$	0.48 ± 0.05	33
$(1.4 \pm 0.5) \times 10^{-10}$	0.91 ± 0.18	this work
$(1.3 \pm 0.2) \times 10^{-10}$	0.87 ± 0.10	this work
$(5.0 \pm 2.2) \times 10^{-13}$	0.98 ± 0.22	32
$(3.1 \pm 1.4) \times 10^{-13}$	0.96 ± 0.24	32
$(1.0 \pm 0.1) \times 10^{-10}$	0.88 ± 0.06	33
$(0.71\pm0.12) \times 10^{-10}$	0.76 ± 0.09	33
	$\begin{array}{c} (2.3\pm0.3)\times10^{-10}\\ (2.4\pm0.5)\times10^{-10}\\ (7.1\pm4.4)\times10^{-11}\\ (1.6\pm0.2)\times10^{-10}\\ (1.4\pm0.1)\times10^{-10}\\ (1.4\pm0.1)\times10^{-10}\\ (1.4\pm0.5)\times10^{-10}\\ (1.3\pm0.2)\times10^{-10}\\ (5.0\pm2.2)\times10^{-13}\\ (3.1\pm1.4)\times10^{-13}\\ (1.0\pm0.1)\times10^{-16}\\ \end{array}$	$\begin{array}{c} \begin{array}{c} (2.3\pm0.3)\times10^{-10} & 1.0\pm0.1 \\ (2.4\pm0.5)\times10^{-10} & 1.1\pm0.1 \\ (7.1\pm4.4)\times10^{-11} & 1.5\pm0.3 \\ (3.3\pm1.4)\times10^{-11} & 1.5\pm0.3 \\ (3.3\pm1.4)\times10^{-11} & 1.4\pm0.1 \\ (1.6\pm0.2)\times10^{-10} & 0.35\pm0.06 \\ (1.4\pm0.1)\times10^{-10} & 0.48\pm0.05 \\ (1.4\pm0.5)\times10^{-10} & 0.87\pm0.10 \\ (5.9\pm2.2)\times10^{-10} & 0.87\pm0.12 \\ (5.9\pm2.2)\times10^{-10} & 0.38\pm0.02 \\ (3.1\pm1.4)\times10^{-13} & 0.96\pm0.22 \\ (3.1\pm1.4)\times10^{-10} & 0.88\pm0.06 \end{array}$

^a Error represents one standard deviation.



Extrapolation uncertainty vs. correlations⁶

	$\ln A$	E_a/R (K)	ρ
Sato <i>et al.</i> (1999)	-22.193 ± 0.13	503 ± 50	N/A



⁶E. Hébrard et al., J. Phys. Chem. A 113:11227-11237 (2009)

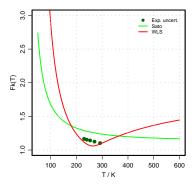
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Extrapolation uncertainty vs. correlations⁷

	$\ln A$	E_a/R (K)	ρ
Hebrard <i>et al.</i> (2009)	-22.193 ± 0.66	503 ± 170	0.996



⁷E. Hébrard et al., J. Phys. Chem. A 113:11227-11237 (2009)

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Extrapolation uncertainty vs. correlations



- I realized later that the $\chi^2 \propto R^2/u^2 >$ value for our best fit was **too small**, meaning that the u_i s were too large to explain the fit residuals
 - it is plausible that systematic errors were aggregated into Sato's uncertainties
- Considering the presence of an unknown constant offset ξ , the statistical model becomes

$$y_i = a + bx_i + \epsilon_i + \xi; \ i = 1, \ N$$

with $E(\epsilon_i) = E(\xi) = 0$ and, noting $Var(\xi) = \sigma_o^2$.

The data are now correlated, with

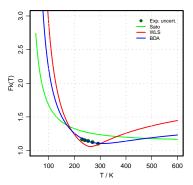
$$cov(y_i, y_j) = cov(\epsilon_i + \xi, \epsilon_j + \xi) = (u_{y_i}^2 - \sigma_o^2)\delta_{ij} + \sigma_o^2$$

+ new fit, with parameters a, b and σ_o (use Bayesian data analysis)



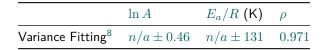
Extrapolation uncertainty vs. correlations

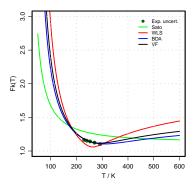
	$\ln A$	E_a/R (K)	ρ	ρ_{sys}
Pernot (later)	-22.16 ± 0.39	530 ± 117	0.967	0.7





Extrapolation uncertainty vs. correlations





⁸Nagy et al. (2012) Rel. Eng. Sys. Saf. 107:29-34 (2012)

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Structural correlations - Multi-pathway react¹CP

rate constants (k) and branching ratios (b_i) (BR) measured in different ۲ experiments

$$\begin{array}{l} \mathbf{I}^{+} + \mathbf{e}^{-} \xrightarrow{k_{1} \pm u_{k_{1}}} \mathbf{M}_{1} \\ \mathbf{I}^{+} + \mathbf{e}^{-} \xrightarrow{k_{2} \pm u_{k_{2}}} \mathbf{M}_{2} \\ \mathbf{I}^{+} + \mathbf{e}^{-} \xrightarrow{k_{3} \pm u_{k_{3}}} \mathbf{M}_{3} \end{array} \qquad \qquad \mathbf{I}^{+} + \mathbf{e}^{-} \xrightarrow{k \pm u_{k}} \begin{cases} \xrightarrow{b_{1} \pm u_{b_{1}}} \mathbf{M}_{1} \\ \xrightarrow{b_{2} \pm u_{b_{2}}} \mathbf{M}_{2}; \\ \xrightarrow{b_{3} \pm u_{b_{3}}} \mathbf{M}_{2}; \end{cases} \sum b_{i} = 1 \end{cases}$$

• Golden Rule of UP: Do not mix separate uncertainty sources!

- essential for sum-to-one b_i
- large impact on uncertainty and sensitivity analysis⁹

⁹Carrasco et al. (2007) J. Phys. Chem. A 111:3507

Effect of ignoring sum-to-one correlation

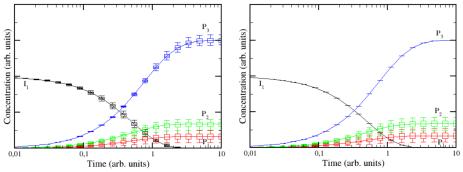
$$I_1 + M_1 \longrightarrow P_1; k_1, b_1$$

$$I_1 + M_1 \longrightarrow P_2; k_1, b_2$$

$$I_1 + M_2 \longrightarrow P_3; k_2$$

with $u(k_1) = u(k_2) = 0$ and $b_1 + b_2 = 1$







Application to dissociative recombination



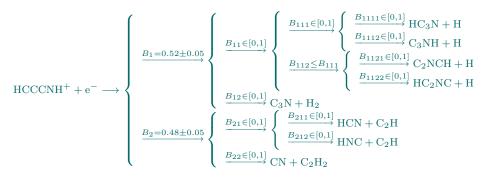
The case of HCCCNH⁺

- Geppert et al. [(2004) ApJ 613:1302-1309] measurements for DCCCND⁺
 - $B_1 = 0.52 \pm 0.05$ for {DC₃N + D; C₃N + D₂}
 - $B_2 = 0.48 \pm 0.05$ for {DCN + C₂D; CN + C₂D₂}
- isotope effects expected to be small: we can transpose information to HCCCNH⁺
- Osamura et al. [(1999) ApJ 519:697-704]
 - interconversion barriers low enough for isomerisation of HC₃N
 - formation of HC₃N more likely than HC₂NC
- HCN / HNC



Application to dissociative recombination

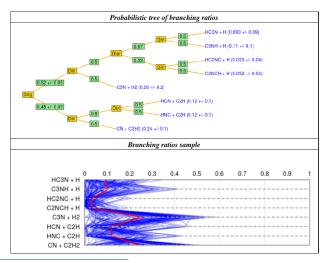
Resulting probabilistic tree



Application to dissociative recombination



Implementation in MC-ChemDB¹⁰



¹⁰(https://github.com/ppernot/MC-ChemDB); Plessis et al. (2010) J. Chem. Phys. 133:134110

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Computational chemistry UQ



- Several methods in 'traditional' CC try to provide prediction uncertainty (Composite methods, Bayesian Ensembles DFT, Statistical corrections...)
 - inexistent to disparate validation methods
- Surge of uncertainty validation needs from Machine Learning (for prediction, or for Active Learning)
 - several validation methods with unclear application limits (interval-based, variance-based, ranking-based)¹¹
- $\bullet\,$ The Calibration/Tightness framework provides a principled validation approach 12

¹¹Scalia *et al.* (2020) *J. Chem. Inf. Model.* (https://tinyurl.com/yc7rn7dp)
¹²Gneiting et al. (2007) *Stat. Meth. B* (https://tinyurl.com/2p8nr3ab); Pernot (2022) *J. Chem. Phys.* (https://tinyurl.com/bdfy8snx); Pernot (2022) (https://arxiv.org/abs/2204.13477)

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Concepts

Calibration / Tightness



Test the statistical compatibility between uncertainties and errors, i.e.

 $E_i \sim D(\mu = 0, \sigma = u_{E_i})$

where D is (often) an unknown distribution.

• (Average) calibration (assuming $\langle E \rangle = 0$)

 $Var(E) = \langle u_E^2 \rangle$, or better, $Var(E/u_E) = 1$

Tightness (small-scale or local calibration):

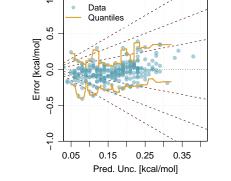
$$\operatorname{Var}\left(E|u_E^2=\sigma^2\right)=\sigma^2,\,\forall\sigma^2$$

¹³Pandey and Jacobsen (2015) Phys. Rev. B (https://tinyurl.com/5dv9spnn), Pernot (2017) J. Chem.

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Formation heats by the mBEEF method¹³

- Bayesian Ensembles method inflates parametric uncertainty of exchange-correlation model to cover errors amplitude
 - strong functional constraints: tightness ???
 - does not disambiguate model uncertainty from reference data uncertainty
- Set of $M = 257 \{V_i, R_i, u_{V_i}\}$





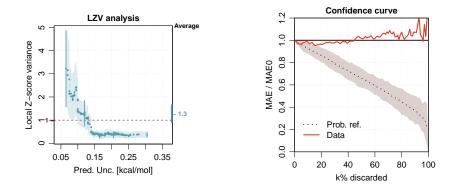
Example

Uncertainty validation E

Example

Formation heats by the mBEEF method



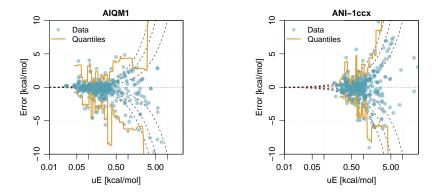


- Var(Z) = 1.3(2), average calibration OK
- *Tightness is bad*: the LZV analysis shows that small PUs are underestimated, while large ones are overestimated
- The confidence curve is catastrophic (cannot use for Active Learning)

Query by Commitee UQ



QbC uncertainties (n = 8) on formation enthalpies for AlQM1 and ANI-1ccx¹⁴. Set of $M = 472 \{E_i, u_{V_i}\}$



¹⁴Zheng et al. (2022) J. Phys. Chem. Lett. (https://tinyurl.com/ccefk79z)

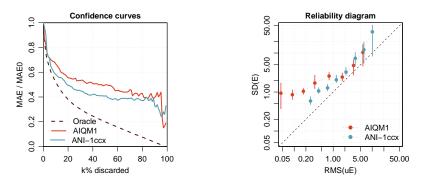
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Query by Commitee UQ

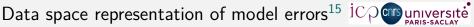


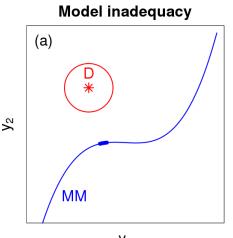
Stat	AIQM1	ANI-1ccx	Target
$\overline{Var(Z)}$	59	4.3	1.4



- QbC does not provide a prediction uncertainty
- but both methods point reliably to largest errors (good for active learning !)

Principles





• Measurement model

 $y_i = M(x_i; \boldsymbol{\vartheta}) + \epsilon_i$

•
$$N_{artheta} < N$$
 (N data points)

• inadequacy: no intersection between the model manifold (MM) and a high probability hypersphere around the data (D)

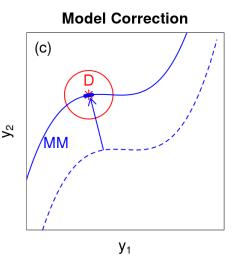
У1

¹⁵Transtrum et al. (2011) Phys. Rev. E 83:036701

Principles



Correction by model discrepancy function¹⁶



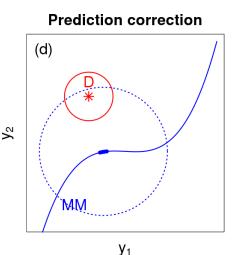
- Measurement model
 - $y_i = \alpha * M(x_i; \boldsymbol{\vartheta}) + \delta(x_i; \boldsymbol{\tau}) + \epsilon_i$
 - $\delta(x_i; \boldsymbol{\tau})$ deterministic (shift) or stochastic (GP, ML...)

Problems

- parameters identifiability
- α et $\delta(x_i; \boldsymbol{\tau})$ not transferable

¹⁶Kennedy, M.C. & O'Hagan, A. (2001) J. R. Stat. Soc. B 63:425.

External stochastic correction¹⁷





• Measurement model

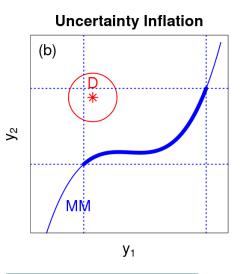
 $y_i = M(x_i; \boldsymbol{\vartheta}) + \boldsymbol{\delta} + \epsilon_i$ $\boldsymbol{\delta} \sim N(0, \sigma)$

Problem

• non-transferability of correction model

¹⁷P. Pernot et al. (2015) J. Phys. Chem. A **119**:5288-5304; Lejaeghere et al. (2014) Crit. Rev. Sol. St. Mat. Sci. **39**:1-24

Stochastic embedding¹⁸





• Measurement model

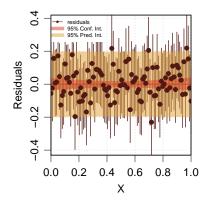
 $y_i = M [x_i; \boldsymbol{\vartheta} + \boldsymbol{\delta}(\boldsymbol{\vartheta})] + \epsilon_i$ $\boldsymbol{\delta} \sim N(0, \Sigma_{\theta})$

- the model parameters become stochastic variables, with a covariance matrix Σ_{θ} adapted for model predictions to **marginally** overlap with the data
- transferability of Σ_{θ} possible, but is it pertinent ???

¹⁸Sargsyan, K. et al. (2015) Int. J. Chem. Kinet. 47:246; Pernot and Cailliez (2017) AIChE J. 63:4642-4665; Pernot (2017) J. Chem. Phys. 147:104102; Sargsyan, K. et al. (2018) arXiv:1801.06768

Linear Calibration

The valid case



- N triplets $\{x_i, y_i, u_{y_i}\}$
- Measurement model (WLS)

 $y_i = a + bx_i + \epsilon_i$ $\epsilon_i \sim N(0, u_{y_i})$

• Confidence and prediction uncertainties at new point x_*

$$\begin{split} y_* &= \hat{a} + \hat{b}x_* \\ u_c^2(y_*) &= u_a^2 + (u_b x_*)^2 + 2 u_a u_b \rho_{ab} x_* \\ u_p^2(y_*) &= u_c^2(y_*) + < u_y^2 > \end{split}$$

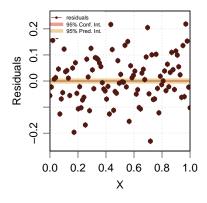
• Note.
$$u_c^2(y_*) \longrightarrow 0$$
 as $N \longrightarrow \infty$
and/or $u_y \longrightarrow 0$



Example

Failure of standard calibration

Model inadequacy

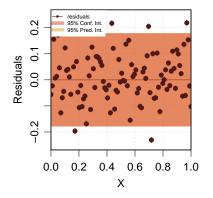


- Same data, with smaller u_{y_i}
- Same measurement model
- The measurement model is inadequate and stastically invalid RMSD >> < u_{yi} >

For inadequate models, one cannot make predictions using the standard calibration setup

Example

Alternative models: Statistical Correction jCpersuniversi



New measurement model

$$y_i = a + bx_i + \delta + \epsilon_i$$

$$\epsilon_i \sim N(0, u_{y_i})$$

$$\delta \sim N(0, \sigma)$$

Parameters: a, b and σ

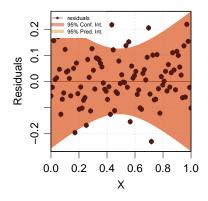
• Confidence and prediction intervals at new point x_{\ast}

$$\begin{split} y_* &= \hat{a} + \hat{b}x_* \\ u_c^2(y_*) &= u_a^2 + (u_b x_*)^2 + 2u_a u_b \rho_{ab} x_* \\ &+ \sigma^2 \\ u_p^2(y_*) &= u_c^2(y_*) + \langle u_y^2 \rangle \end{split}$$

• Problem: non-transferability of σ

Example





• Standard measurement model with scaled parameters covariance matrix

$$u_c^{\prime 2}(y_*) = T * [u_a^2 + (u_b x_*)^2 + 2u_a u_b \rho_{ab} x_*]$$

i.e.,

$$\Sigma'_\vartheta = T * \Sigma_\vartheta$$
 with T such as $< u_c'^2(y_*) > \simeq \mathrm{Var}(R)$

- Advantage: transferability of $\Sigma'_{\mathfrak{A}}$ (?)
- Problems
 - confidence intervals geometry is determined by model's jacobian
 - uncontrolled risk of local under/overestimation

¹⁹Pernot and Cailliez (2017) AIChE J. 63:4642-4665; Pernot (2017) J. Chem. Phys. 147:104102

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Conclusions



- Probabilistic uncertainty representation
 - use adapted pdfs that implement all physical constraints
 - Correlation is your friend !
- Uncertainty validation
 - Calibration is easy, tightness is hard !
 - Check calibration and tightness with appropriate methods
- Uncertainty estimation with inadequate models
 - Validate the statistical model
 - Simplest ideas are not necessarily the best ones !